

## THE ABSORPTION SPECTRUM AND DICHROISM OF POTASSIUM TETRACHLOROPLATINATE(II) CRYSTALS AT 15 °K.

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The absorption spectra of thin crystals of  $K_2PtCl_4$  with plane-polarized light at room temperature were presented to the 1964 Bressanone Conference (see p. 39). Discussions at that conference emphasized the uncertainties in the assignments of transitions in the spectra and the desirability for measurements at low temperatures where thermal excitation of molecular vibrations would not be significant. We are presently reporting the spectra obtained in the past few months with single crystals maintained at a nominal temperature of 15 °K in a liquid helium cryostat.

It is, of course, well established that the  $PtCl_4^{2-}$  ion has a square-planar arrangement of ligands with a  $5d^8$ -electronic configuration. The arrangement of energy levels depends upon the degree of tetragonal distortion which is superimposed upon the octahedral splitting of the  $d$ -orbitals. Fig. 1 shows three possible arrangements of the  $d$ -orbitals for  $PtCl_4^{2-}$  as an increasing tetragonal distortion is applied to the familiar  $O_h$  arrangement of orbitals. Various workers<sup>1,2,3,4</sup> have proposed each of these arrangements in assigning the transitions in the spectra.

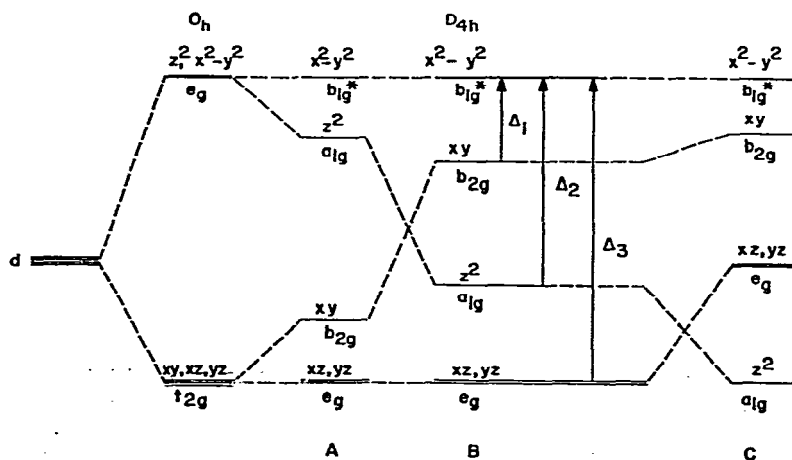


Fig. 1. Possible ordering of the  $5d$  orbital energies as a tetragonal distortion is applied to an octahedral coordination.

Also, since platinum as a heavy element has a moderately high spin-orbit coupling, the possibility of spin-forbidden transitions exists. In solution and crystal spectra two strong  $d-d$  bands and some much weaker bands are observed which clearly are spin-allowed and spin-forbidden respectively. However, there is one band observed which is intermediate in intensity between the two groups and its presence has contributed an additional uncertainty in the assignment of transition states. At present, molecular orbital calculations have not reached a degree of refinement where these ambiguities can be resolved by theoretical means.

The new data are presented as Figs. 2 and 3 where molar extinction coefficients for the crystals with light polarized in the  $z$  direction ( $\parallel$  to the symmetry axis) and in the  $x, y$  direction ( $\perp$  to the symmetry axis) are plotted for observations at  $298^\circ$  and  $15^\circ K$ . In addition, the unpolarized absorption of a crystal 0.55 mm thick was recorded, and  $\epsilon$  was  $<0.01 \text{ cm}^{-1} M^{-1}$  for the region  $8,000$  to  $14,300 \text{ cm}^{-1}$ .

It is apparent that the intensity of each major transition was reduced by a factor of approximately .3 to .5 at  $15^\circ K$ . In addition, the bands were significantly narrower and the maxima had shifted to higher energies at the lower temperatures.

One of the most striking features of the low temperature spectra was the resolution of vibrational structure in the absorption bands over the region  $22,300$  to  $27,000 \text{ cm}^{-1}$  for the  $x, y$  polarization. Some 18 maxima can be resolved in this region. In addition, a weak transition band at about  $24,000 \text{ cm}^{-1}$  in  $x, y$ -polarization is now clearly evident. In retrospect, it can be seen that this transition corresponds in energy to a very weak absorption which is discernible in the  $z$ -polarization of the room temperature spectra but it fell under the edge of the band with maximum at  $25,700 \text{ cm}^{-1}$  in  $x, y$  polarization. In contrast to the above, not the slightest structure could be discerned in the largest absorption band in the region of  $28,000$ – $31,000 \text{ cm}^{-1}$  for either polarization.

Fig. 3 presents spectra for a thicker crystal which was used to give higher absorption intensities in the long wavelength region. In these curves structure is clearly evident in the  $z$ - or  $\parallel$ -polarization peak near  $20,600 \text{ cm}^{-1}$ , although it is much less pronounced than in regions above  $23,000 \text{ cm}^{-1}$ . For the  $x, y$ - or  $\perp$ -polarization the structure was still fainter and only barely evident. The transition at  $24,000 \text{ cm}^{-1}$  in  $z$ -polarization is clearly evident for this crystal and in consideration of the weakness of the transition exhibits a strong structuring. It is clear that there is a weak absorption band centered at  $18,000 \text{ cm}^{-1}$  in  $x, y$ - or  $\perp$ -polarization. Fine structure was weakly evident in this region. The absorption in  $z$ -polarization at this wavelength is much weaker still; however, there is a greater intensity of absorption increasing from  $15,000$ – $19,000 \text{ cm}^{-1}$  than corresponds to the tail of a Gaussian peak centered about  $20,600 \text{ cm}^{-1}$ .

The energies of the transitions indicated from Figs. 2 and 3 have been included in Table I together with the extinction coefficients for the bands at  $15^\circ K$ .

The  $d^8$  configuration can be treated by the two-hole formalism. Wave functions are expressed as products of the orbital and spin functions and the symmetry

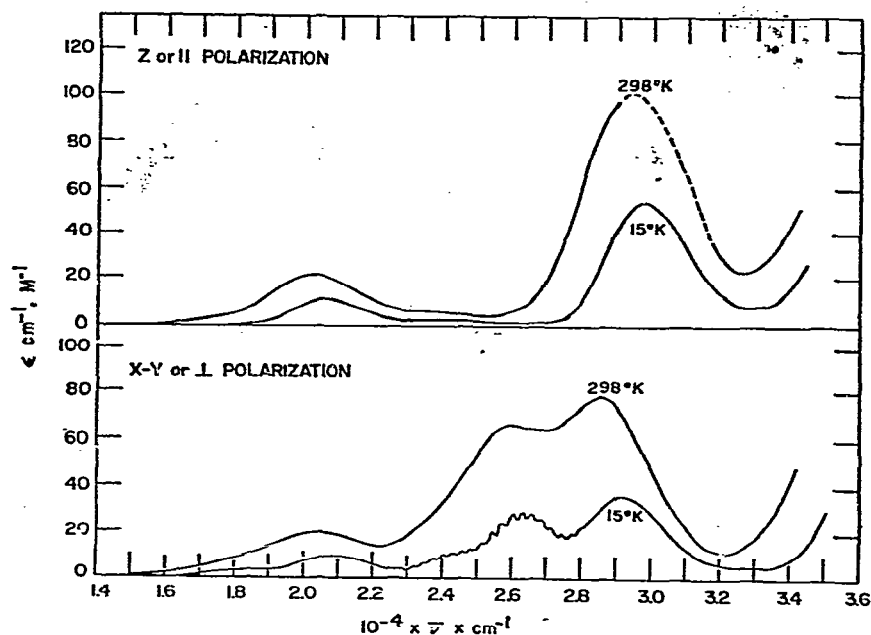


Fig. 2. Absorption spectra of a  $K_2PtCl_6$  crystal with polarized light. Crystal thickness = 46.5 microns.

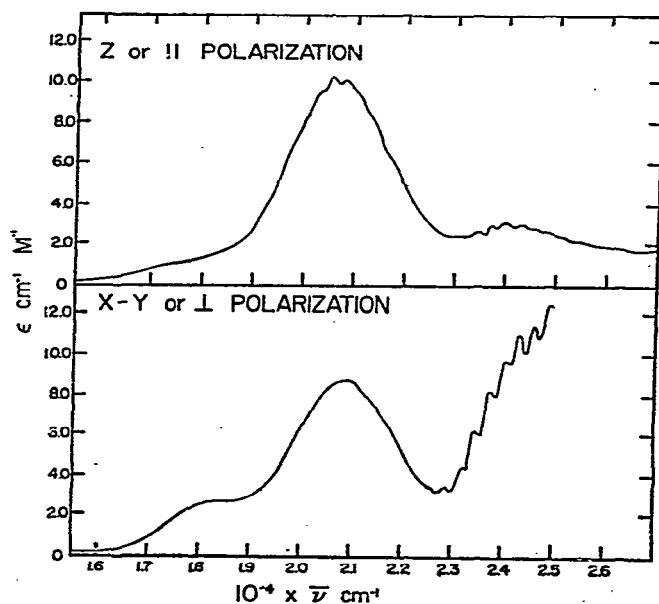


Fig. 3. Absorption spectra of a  $K_2PtCl_6$  crystal with polarized light at 15 °K. Crystal thickness = 113 microns.

TABLE I

OBSERVED ABSORPTION PEAKS AT 15 °K FOR  $K_2PtCl_4$  AND CALCULATED ENERGY LEVELS

<i>z</i> -Polarization		<i>x, y</i> -Polarization		Vibrational Fine Structure	Calculated Energy ( $cm^{-1}$ )	Principal Component
$\bar{\nu}$ ( $cm^{-1}$ )	$\epsilon$ ( $M^{-1}cm^{-1}$ )	$\bar{\nu}$ ( $cm^{-1}$ )	$\epsilon$ ( $M^{-1}cm^{-1}$ )			
17,000	<1				$\Gamma_1 - 16,100$	$^3(b_{1g}e_g)$
19,000				none	$\Gamma_5 - 18,700$	$^3(b_{1g}e_g)$
		18,000	2	very weak	$\Gamma_2 - 17,800$	$^3(b_{1g}e_g)$
20,600	10	20,900	8.7	weak <i>z</i>	$\Gamma_4 - 20,800$	$^3(b_{1g}e_g)$
				very weak	$\Gamma_5 - 21,100$	$^3(b_{1g}b_{1g})$
				<i>xy</i>	$\Gamma_3 - 21,300$	$^3(b_{1g}e_g)$
24,000	3	24,100	7.1	strong	$\Gamma_1 - 23,500$	$^3(b_{1g}b_{2g})$
		26,300	28	strong	$\Gamma_2 - 26,300$	$^1(b_{1g}b_{2g})$
29,800	55	29,200	37	none	$\Gamma_5 - 29,400$	$^1(b_{1g}e_g)$
					$\Gamma_4 - 36,100$	$^3(b_{1g}a_{1g})$
					$\Gamma_5 - 36,400$	$^3(b_{1g}a_{1g})$
					$\Gamma_3 - 46,285$	$^1(b_{1g}a_{1g})$

*d*-state transitions above  $34,000\text{ cm}^{-1}$  are hidden under the edge of the intense allowed charge transfer band.

properties of the total wave function determined by the rules of group theory for direct products. Thus the singlet spin function transforms as  $a_{1g}$  and the triplet functions as  $a_{2g}$  and  $e_g$  in the  $D_{4h}$  group. The triplet states are therefore split as a consequence of the spin orbit coupling.

Bethe's  $\Gamma$ -notation is used for describing the total spin-orbital wave-functions, i.e.:  $\Gamma_1 = a_{1g}$ ,  $\Gamma_2 = a_{2g}$ ,  $\Gamma_3 = b_{1g}$ ,  $\Gamma_4 = b_{2g}$  and  $\Gamma_5 = e_g$ . In addition, the *x* and *y* axes have been chosen to pass through the ligand positions.

The energy levels of the  $PtCl_4^{2-}$  were calculated including the contributions of one-electron orbital energies (the  $\Delta_1$ ,  $\Delta_2$ ,  $\Delta_3$  in Fig. 1), the electron-electron repulsions (the Slater-Condon parameters  $F_2$  and  $F_4$ ) and spin-orbit coupling.

The set of parameters which gave most satisfactory agreement with energies and intensities of the observed bands is given in Table II.

The calculated energies together with the symmetry representations and the principle contributing state for each level are also presented in Table I. The  $d_{z^2}$  or  $a_{1g}$  orbital has the lowest energy in this scheme and has arbitrarily been assigned zero energy.

TABLE II

PARAMETERS UTILIZED IN CALCULATING THE ENERGY LEVEL SCHEME IN TABLE I

Parameter	Energy ( $cm^{-1}$ )
$d_{z^2}$ orbital ( $a_{1g}$ )	0
$d_{xz, yz}$ orbitals ( $e_g$ )	17,300
$d_{xy}$ orbital ( $b_{2g}$ )	24,700
$d_{x^2-y^2}$ orbital ( $b_{1g}$ )	51,000
$F_2$	1,200
$F_4$	65
Spin-Orbit Coupling $\alpha = \zeta/2$	1,300

The energy levels for the system have been plotted as a function of the spin-orbit coupling parameter in Fig. 4 to illustrate the importance of this quantity. The squares in the figure indicated the observed band maxima. Thus, the states originating from the two triplet states of nearly equal energy in the absence of spin-orbit coupling split by as much as  $8,000 \text{ cm}^{-1}$ . The band of intermediate intensity at  $ca. 20,700 \text{ cm}^{-1}$  arises from transitions to the four  $\Gamma_3$ ,  $\Gamma_4$  and  $\Gamma_5$  states.

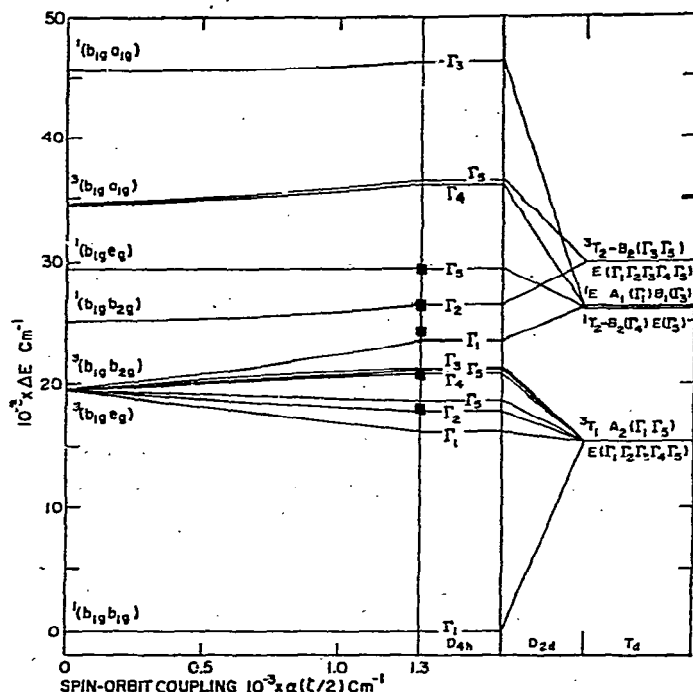


Fig. 4. The splitting of the square-planar electronic states by spin-orbit coupling with their  $D_{4h}$ - $\Gamma$ -designation and the correlation of the square-planar states with the tetrahedral states (without spin-orbit coupling). The representations of the orbital part of the tetrahedral states under the  $D_{2d}$  intermediate symmetry group are shown together with the  $\Gamma$ -designations, of the total wave functions under  $D_{2d}$ .  $F_2 = 1200$ ,  $F_4 = 65$ ,  $A_1 = 26,300$ ,  $A_2 = 51,000$ ,  $A_3 = 33,700$  and  $\Delta_1 \approx 17,000 \text{ cm}^{-1}$ .

It is necessary to have a theoretical model which will account for the intensities of the observed spectra. According to the simple vibronic model, asymmetric vibrations provide a perturbation which mixes an asymmetric component into the wave functions so that a non-zero dipole transition moment can be obtained. From the symmetry properties of the normal vibrational coordinates for  $\text{PtCl}_4^{2-}$ , the simple vibronic model predicts that transitions to  $\Gamma_2$  and  $\Gamma_4$  states will be polarized in the  $x$ ,  $y$ -direction which is consistent with the assignment of the peak at  $26,300 \text{ cm}^{-1}$ . The moderate absorption at  $18,000 \text{ cm}^{-1}$  is also assigned as a  $\Gamma_2$  although there is a much weaker absorption in this region which is attributed to the  $\Gamma_1$  at

16,100 and the  $\Gamma_5$  at 18,700  $cm^{-1}$ . Under favorable circumstances, where the potential functions for all the vibrations of an excited electronic state except the totally symmetric stretching vibration have similar force constants and equilibrium positions as the corresponding vibrations in the ground state, the vibronic model predicts a strong vibrational structure, as was observed for the transitions at 24,000 and 26,300  $cm^{-1}$ . The vibrational structure can be attributed to a series of vibrational progressions in the symmetric stretching mode. Thus if the equilibrium distance for the Pt-Cl bonds is greater in the excited electronic state, the Frank-Condon principle requires that there will be a maximum probability for the ion to be carried into a highly excited  $Q_1$ -vibration state. The separations of the maxima in these bands amount to 270–290  $cm^{-1}$  which, as expected, are somewhat lower than the  $Q_1$ -frequency of 335  $cm^{-1}$  which is available from Raman spectra.

The out-of-plane  $\beta_{2u}$  bending vibration carries the square-planar,  $D_{4h}$ , arrangement into the symmetry  $D_{2d}$ . If continued, this vibration carries the ion into a tetrahedral,  $T_d$  arrangement. The ground state for  $T_d$  symmetry, without spin-orbit coupling, is  $^3T_1$ . It is proposed that the tetrahedral ground state lies below some of the excited states for the square-planar arrangement. The nine wave functions which comprise the  $^3T_1$  in the  $T_d$  arrangement must each correlate via basis functions for the  $D_{2d}$  symmetry group of the intermediate configuration with states in the square-planar arrangement. If the  $^3T_1$  state is below the  $D_{4h}$  state with which it correlates, the potential functions for the  $\beta_{2u}$  out-of-plane bending vibration for the  $D_{4h}$  ion will possess one or two maxima. Transitions to a number of vibration levels, which are not uniformly spaced, in the excited electronic state are to be expected; and the vibrational fine structure of the absorption bands may be lost. All of the  $D_{4h}$  states up to and including the  $\Gamma_3$  at 21,300  $cm^{-1}$  (Table I) correlate with the  $^3T_1$  ground state of  $T_d$ .

The correlation of the  $D_{4h}$  states with the  $T_d$  states is also included in Fig. 4. For the  $T_d$  states, zero spin-orbit coupling was assumed and the values of  $F_2$  and  $F_4$  from the  $D_{4h}$  system were employed. The parameter,  $\Delta_t$ , was set equal to 17,000  $cm^{-1}$  which is somewhat greater than the crystal field value of  $4/9\Delta_1$ . The ground state of  $T_d$  was placed just below the first excited  $\Gamma_1$ -state of  $D_{4h}$ . The  $\Gamma_1$  at 23,500  $cm^{-1}$  and the  $\Gamma_2$  state at 26,400  $cm^{-1}$  correlate with  $T_d$  states of higher energy. These are just the ones reached by transitions with strong vibrational structure. The  $\Gamma_5$  state at 29,400  $cm^{-1}$  logically correlates with a state of lower energy so that the absence of structure can be explained.

One additional piece of evidence can now be cited for the assignment of the 29,400  $cm^{-1}$  transition to  $\Gamma_5$  states. Very recently Professor J. G. Foss of the Iowa State University Department of Biochemistry-Biophysics has recorded a magnetic circular dichroism which is associated with the solution absorption band at 30,300  $cm^{-1}$ . The shape and magnitude of this dichroism indicates that it originates from the Zeeman splitting of a degenerate state. Such a splitting would require a state with the  $\Gamma_5$ -symmetry in the  $D_{4h}$  group.

A number of other transition assignments have been considered, but it is believed that the scheme included in Table I, with the modifications in the vibronic model which are required for the tetrahedrally deformed excited states, provides the most satisfactory explanation of the energies and intensities in the observed spectra.

#### *Acknowledgement*

Figs. 1, 2, 3 and 4 are reproduced, with permission, from D. S. MARTIN *et al.*, *Inorg. Chem.*, 4 (1965) 1682.

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